

Silicon Nanoparticles prepared via Laser Based Green Synthesis for energy and biomedical applications

Romuald Intartaglia, Komal Bagga, Fernando Brandi and Alberto Diaspro

Department of Nanophysics, Istituto Italiano di Tecnologia, via Morego, 30, 16163 Genova, Italy

E-mail : romuad.intartaglia@iit.it

Abstract:

During the last decade, silicon nanoparticles (Si-NPs) have triggered large research activity due to their particular size-dependent optical properties for the silicon based photonic technology. Recently, Si-NPs are emerging as promising nanostructures for potentials applications in biomedical field such as in vivo fluorescence imaging label¹, bio imaging contrast agent² and therapy³. Up to now, a variety of method have been developed for the Si-NPs synthesis.

In this work, we report the synthesis of NPs via Pulsed Laser Ablation in Liquid (PLAL). PLAL have emerged as an appealing alternative approach to generate pure metallic⁴ and semiconductor⁵ NPs. This versatile method, presents several advantages compared with the chemical method, i) Preparation is simple and clean since the synthesis can be carried out in water or in solution of a biocompatible ligand ii) Direct solubility of the product in polar solvent due to unique NPs surface chemistry. Si-NPs is produced by femtosecond pulsed laser ablation of silicon target in deionized water. The obtained colloidal solution are characterized by different techniques : Optical spectroscopy, microRaman spectroscopy, HRTEM and STEM measurements. Size control of Si-NPs via PLAL is obtained changing the laser pulse energy. Figure 1 shows STEM and HRTEM image of the obtained Si-NPs at lower laser pulse energy. Moreover, effects of the liquid environment revealed to play a key role on the microstructure of the obtained Si-NPs and on the optical properties of the colloidal solution.

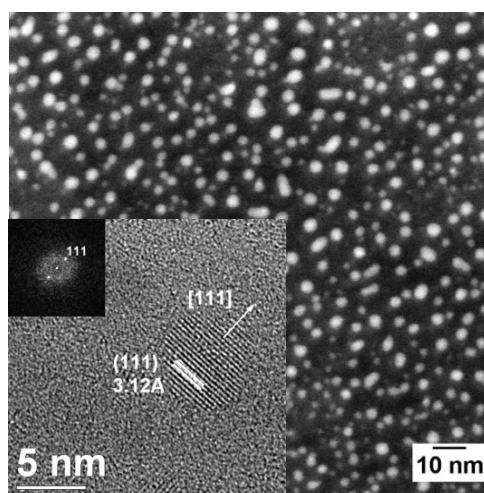


Figure : STEM image of ultra small Si-NPs prepared in deionized water via laser based method. In inset is reported HRTEM image of a single synthesized Si-NP showing the (111) lattice sets and its corresponding numerical electron diffraction pattern.

References:

- [1] Erogbogbo, F.; Yong, K.T.; Roy, I.; Hu, R.; Law, W.C.; Zhao, W.; Ding, H.; Wu, F.; Kumar, R.; Swihart, M.T.; Prasad, P.N. *ACS Nano*. **2011**, 5 (1), 413-423.
- [2] Apteekar, J.W.; Cassidy, M.C.; Johnson, A.C.; Barton, R.A.; Lee, M.; Ogier, A.C.; Vo, C.; Anahtar, M.N.; Ren, Y.; Bhatia, S.N.; Ramanathan, C.; Cory, D.G.; Hill, A.L.; Mair, R.W.; Rosen, M.S.; Walsworth, R.L.; Marcus, C.M. *ACS Nano*. **2009**, 3 (12), 4003-4008.
- [3] Xiao, L.; Gu, L.; Howell, S.B.; Sailor, M.J. *ACS Nano*. **2011**, 5, 3651-3659.
- [4] Petersen, S.; Barcikowski S. *Adv. Funct. Mater.* **2009**, 19, 1167-1172
- [5] Intartaglia, R.; Bagga, K.; Brandi, F.; Genovese, A.; Di Fabrizio, E.; Diaspro, A. *J. Phys. Chem. C*. **2011**, 115, 5102-5107.